

WEEKLY TEST TYJ -1 TEST - 33 R SOLUTION Date 29-12-2019

[PHYSICS]

16.

1. (a)
$$v = n\lambda = 2 \times 5 = 10 \text{ cm/sec}$$

2. (a)
$$v_{\text{max}} = a\omega = a \times 2\pi n = 0.1 \times 2\pi \times 300 = 60\pi \text{ cm} / \text{sec}$$

3. (c) Phase difference
$$=\frac{2\pi}{\lambda}\times$$
 path difference $\Rightarrow 1.6\pi = \frac{2\pi}{\lambda}\times 40 \Rightarrow \lambda = 50 \ cm = 0.5m$ $\Rightarrow v = n\lambda \Rightarrow 330 = 0.5\times n \Rightarrow n = 660 \ Hz$

5. (a)
$$\lambda = \frac{v}{n} = \frac{1.7 \times 1000}{4.2 \times 10^6} = 4 \times 10^{-4} \text{ m}$$

6. (c) Velocity of sound in gas
$$v = \sqrt{\frac{\gamma RT}{M}} \Rightarrow v \propto \sqrt{\frac{\gamma T}{M}}$$

$$\Rightarrow \frac{v_{N_2}}{v_{He}} = \sqrt{\frac{\gamma_{N_2}}{\gamma_{He}} \times \frac{M_{He}}{M_{H_2}}} = \sqrt{\frac{\frac{7}{5}R \times 4}{\frac{5}{3}R \times 28}} = \frac{\sqrt{3}}{5}$$

- 7. (a) Time required for a point to move from maximum displacement to zero displacement is $t = \frac{T}{4} = \frac{1}{4n}$ $\Rightarrow n = \frac{1}{4t} = \frac{1}{4 \times 0.170} = 1.47 \, Hz$
- **8.** (c) $\lambda = \frac{v}{n} = \frac{340}{200} = 1.7 \text{ m}$
- **9.** (a) The time taken by the stone to reach the lake

$$t_1 = \sqrt{\frac{2h}{g}} = \sqrt{\frac{2 \times 500}{10}} = 10 \text{sec}$$
 (Using

$$h = ut + \frac{1}{2}gt^2$$

Now time taken by sound from lake to the man

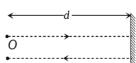
$$t_2 = \frac{h}{v} = \frac{500}{340} \approx 1.5 \text{ sec}$$

 \Rightarrow Total time = $t_1 + t_2 = 10 + 1.5 = 11.5 \text{ sec.}$

10. (b) Distance between a compression and the nearest rarefaction is $\frac{\lambda}{2} = 1m$. Hence $n = \frac{v}{\lambda} = \frac{360}{2} = 180 \, Hz$.

11. (a)
$$v = \sqrt{\frac{\gamma P}{\rho}} \Rightarrow \frac{v_{O_2}}{v_{H_2}} = \sqrt{\frac{\rho_{H_2}}{\rho_{O_2}}} = \sqrt{\frac{1}{16}} = \frac{1}{4}$$

- **12.** (d) Speed of sound in gases is $v = \sqrt{\frac{\gamma RT}{M}} \Rightarrow T \propto M$ (Because v, γ -constant). Hence $\frac{T_{H_2}}{T_{O_2}} = \frac{M_{H_2}}{M_{O_2}}$ $\Rightarrow \frac{T_{H_2}}{(273 + 100)} = \frac{2}{32} \Rightarrow T_{H_2} = 23.2K = -249.7^{\circ}C$
- **13.** (c) Path difference $\Delta = \frac{\lambda}{2\pi} \times \phi \implies 1 = \frac{\lambda}{2\pi} \times \frac{\pi}{2} \implies \lambda = 4m$ Hence $v = n\lambda = 120 \times 4 = 480 \ m/s$
- **14.** (a) Suppose the distance between shooter and reflecting surface is *d*. Hence time interval for hearing echo is



$$t = \frac{2d}{v} \implies 8 = \frac{2d}{350} \implies d = 1400 \, m \, .$$

15. (d) $v = \sqrt{\frac{\gamma P}{\rho}}$; as *P* changes, ρ also changes. Hence $\frac{P}{\rho}$

remains constant so speed remains constant.

(b)
$$d_1 \longrightarrow d_2 \longrightarrow$$

$$2d_1 + 2d_2 = v \times t_1 + v \times t_2 \Rightarrow 2(d_1 + d_2) = v(t_1 + t_2)$$
$$d_1 + d_2 = \frac{v(t_1 + t_2)}{2} = \frac{340 \times (1.5 + 3.5)}{2} = 850 \text{ m}.$$

17. (b) By using
$$v = \sqrt{\frac{\gamma RT}{M}} \Rightarrow v \propto \sqrt{T}$$

$$\frac{v_2}{v_1} = \sqrt{\frac{T_2}{T_1}} = \sqrt{\frac{T + 600}{T}} = \sqrt{3} \Rightarrow T = 300 \, K = 27^{\circ} \, C$$

18. (c)
$$v = \sqrt{\frac{\gamma RT}{M}} \implies v \propto \sqrt{T}$$

i.e. if v is doubled then T becomes four times,

hence $T_2 = 4T_1 = 4(273 + 27) = 1200K = 927^{\circ}C$

- **19.** (c) Since solid has both the properties (rigidity and elasticity)
- **20.** (b) Frequency of wave is $n = \frac{3600}{2 \times 60} Hz \Rightarrow$

$$\lambda = \frac{v}{n} = \frac{760}{30} = 25.3 \text{ m}.$$

[CHEMISTRY]

22.

Reactivity of alkali metals: Li < Na < K < Rb < Cs.

Reactivity of halogens: Fe > Cl > Br > I

23.

It is as per their emf values. The reducing power does not only depend upon ionisation energy but also on enthalpy of atomisation and enthalpy of hydration also.

24 A

25.

The maximum covalency of Be is 4, e.g., $Na_2[Be(OH)_4]$ while that of Al is 6, e.g., $Na_3[AlF_6]$.

26. A

27. E

28.

IE of Mg: Na < Al < Mg

29.

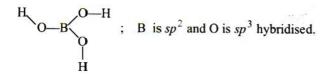
$$KO_2: +1+2x=0 \implies x=-\frac{1}{2}$$

30.

$$B(Z = 5): 1s^2 2s^2 2p^1$$
; $AI(Z = 13): 1s^2 2s^2 2p^6 3s^2 3p^1$

31. A

32.



33. D

34.

$$\frac{1}{2}\text{Al}_2\text{Cl}_6 + 6\text{H}_2\text{O} \longrightarrow [\text{Al}(\text{H}_2\text{O})_6]^{3+} + 3\text{Cl}^{-}$$

- Cs because of its low IE emits electron under the influence of even candle light.
- 36. LiHCO₃ is unstable and exists only in solution.
- Non-metal oxides being acidic decompose carbonates to evolve CO₂ gas.
- 38. NaNO₃ is called chile salt petre.
- 39. Mixture of K₂CO₃ and Na₂CO₃ is called fusion
- 40. BaCO₃ > SrCO₃ > CaCO₃ > MgCO₃. Thermal stability decreases as the basic character of the metal hydroxide decreases.

[MATHEMATICS]

- 1. (a) Required probability = $\frac{4}{7}$.
- **2.** (c) Required probability $=\frac{4}{36} = \frac{1}{9}$.
- **3.** (a) Required probability $=\frac{4}{52}=\frac{1}{13}$.
- **4.** c) Since there are one A, two I and one O, hence the required probability $=\frac{1+2+1}{11}=\frac{4}{11}$.
- **5.** (b) Required probability is 1-P (All letters in right envelope) $=1-\frac{1}{n!}$

{As there are total number of n! ways in which letters can take envelopes and just one way in which they have corresponding envelopes}.

6. (a) Favourable ways {29,92,38,83,47,74,56,65}

Hence required probability $=\frac{8}{100} = \frac{2}{25}$.

- 7. (c) To be both boys the probability $= \left(\frac{1}{2}\right) \left(\frac{1}{2}\right) = \frac{1}{4}$.
- **8.** (b) Let P(A) and P(B) be the probability of the events then $P(A \text{ and } B) = P(A) \cdot P(B) = \frac{1}{2} \times \frac{2}{3} = \frac{1}{3}$.
- 9. (c) Total rusted items =3+5=8; unrusted nails =3.

 $\therefore Required probability = \frac{3+8}{6+10} = \frac{11}{16}$

- 10. (b) Required probability $=\frac{2+1}{36}=\frac{1}{12}$.
- **11.** (b) Total number of ways = 36

Favourable numbers of cases are $(1 \ 4), (2, \ 3), (3, \ 2), (4, \ 1), (1, \ 5), (2, \ 4), (3, \ 3), (4, \ 2), (5, \ 1) = 9$

Hence the required probability $=\frac{9}{36} = \frac{1}{4}$.

12. (c) Total ways are 8 and favourable ways are 4 $S = \{HHH, HHT, ..., TTT\}$.

Hence probability $=\frac{4}{8} = \frac{1}{2}$.

- 13. (b) In a non-leap year, we have 365 days *i.e.*, 52 weeks and one day. So, we may have any day of seven days. Therefore, 53 Sunday, required probability = $\frac{1}{7}$.
- **14.** (a) Required probability $=\frac{5}{25} = \frac{1}{5}$.
- **15.** (d) Required probability $=\frac{4.4}{52.51} \times 2 = \frac{8}{663}$
- **16.** (d) Probability for white ball $P(W) = \frac{4}{15}$

Probability for red ball $P(R) = \frac{6}{15}$

Probability (white or red ball) = P(W) + P(R)

$$=\frac{4}{15}+\frac{6}{15}=\frac{10}{15}=\frac{2}{3}.$$

- 17. (d) Required probability = $\frac{64}{64}$.
- **18.** (c) P (tail in 3^{rd}) P (tail in 4^{th}) $=\frac{1}{2}\cdot\frac{1}{2}=\frac{1}{4}$.
- 19. (b) It is obvious.
- 20. (b) Same number can appear in 6 ways.

Hence required probability $=\frac{6}{216} = \frac{1}{36}$